Experimental Quantum Simulation of Chemical Dynamics

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Simulating chemistry is likely to be among the earliest applications of quantum computing. However, existing digital quantum algorithms for chemical simulation require many logical qubits and gates, placing practical applications beyond existing technology. Here, we use an analog approach to carry out the first quantum simulations of chemical reactions. In particular, we simulate photoinduced non-adiabatic dynamics, one of the most challenging classes of problems in quantum chemistry because they involve strong coupling and entanglement between electronic and nuclear motions. We use a mixed-qudit-boson (MQB) analog simulator, which encodes information in both the electronic and vibrational degrees of freedom of a trapped ion. We demonstrate its programmability and versatility by simulating the dynamics of three different molecules as well as open-system dynamics in the condensed phase, all with the same quantum resources. Our approach requires orders of magnitude fewer resources than equivalent digital quantum simulations, demonstrating the potential of analog quantum simulators for near-term simulations of complex chemical reactions.

Using quantum machines for simulation of nature is the idea that launched quantum computing [1], because it was shown that a controllable quantum device could mimic another quantum system efficiently (with polynomial resource scaling) [2]. Simulations of quantum chemistry, encompassing both electronic structure and dynamics, are particularly well suited for quantum simulation and are likely to be the earliest applications of quantum computing [3–5]. Most proposed chemical applications of quantum computing have focused on finding static molecular properties, usually energies [3, 4, 6– 11]. By contrast, there are few quantum algorithms for molecular dynamics [12–15], despite the centrality of dynamics to chemical reactivity and kinetics. Useful applications of algorithms for chemical dynamics would require digital quantum computers with many more low-error qubits and gates than is feasible with existing technology [12, 14, 15].

A significant opportunity for quantum simulation is the simulation of photochemical reactions, which are both important and difficult to simulate classically. They describe some of the most fundamentally and practically important chemical phenomena, including almost all chemical reactions in the atmosphere, such as those responsible for smog formation and ozone depletion; interactions of organisms with light, whether through photosynthesis, vision, or UV DNA damage; solar energy conversion through solar cells; and compounds for preventing and treating disease, from sunscreens to photodynamic therapies [16]. Simulating photoexcited dynamics is one of the most challenging problems in quantum chemistry, putting the prediction of many reaction properties out of reach. The difficulty arises because these processes are almost invariably non-adiabatic, involving the breakdown of the Born-Oppenheimer approximation [17]. Non-adiabatic processes involve the entangled motion of nuclear wavepackets on multiple potentialenergy surfaces on ultrafast timescales [17, 18]. Methods to simulate non-adiabatic dynamics on conventional

computers are either approximate or limited by computational scaling that is unpredictable a priori [17, 19]. Even more complicated is simulating open quantum systems, including chemical dynamics in the condensed phase [20]. On classical computers, doing so usually involves tracking density matrices (more complicated than wavefunctions) as well as the system's interactions with its environment. Classical multi-configuration time-dependent Hartree (MCTDH) calculations have been achieved for relatively small systems with explicit bath modes [19, 21, 22] or weak system-bath couplings [23], but increasing the coupling strength or the number of modes quickly becomes intractable.

Analog quantum simulation provides a more efficient pathway to solving quantum chemical-dynamics problems [24–31]. An analog simulator is a purpose-built device whose Hamiltonian can be tuned to realize a mapping of the target Hamiltonian to be simulated, without the high qubit and gate overheads of digital approaches. In particular, mixed-qudit-boson (MQB) analog simulators allow simulations of molecular models with vibrational-electronic (vibronic) couplings in a hardware-efficient fashion [24, 31]. This is achieved using qudits (d-level quantum systems) controllably coupled to bosonic modes to encode, respectively, molecular electronic states and nuclear vibrations (Fig. 1). MQB simulators based on trapped-ion systems have been used to predict vibronic spectra and to observe the dynamical geometric-phase interference around an engineered conical intersection [25–27, 32]. These previous experimental analog simulations of chemical dynamics were important but limited demonstrations, restricted to non-molecular model Hamiltonians or being non-programmable. A programmable MQB simulator allows tunable experimental parameters to simulate a wide range of molecules using the same hardware.

Here, we perform the first programmable MQB simulations of non-adiabatic photochemical dynamics using a trapped-ion system. We demonstrate the programmabil-

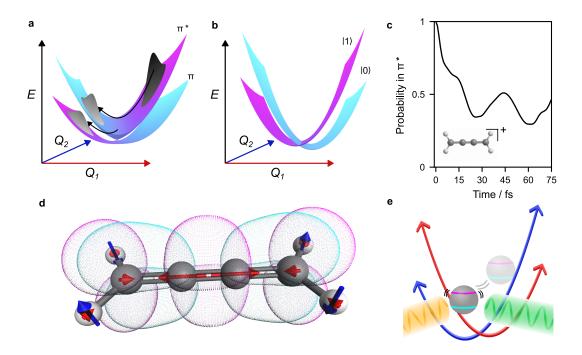


Figure 1. Mapping non-adiabatic chemical dynamics onto an MQB simulator. (a) Adiabatic potential energy surfaces (PES) of the butatriene cation determined from solving the Born-Oppenheimer Hamiltonian at all nuclear coordinates. The cyan and magenta regions of the surfaces correspond to π and π^* electronic characters. The two PESs intersect at a conical intersection (CI) through a coupling with the central bond stretching (Q_1) and torsion (Q_2) vibrational modes. This vibronic coupling allows ultrafast molecular dynamics: an initial wavepacket (grey, top right) on the upper surface approaches the conical intersection and splits into two entangled branches on the two electronic surfaces. (b) Diabatic representation of the PESs. The coupling between surfaces (not shown) is linear along Q_2 . (c) Probability of the wavepacket remaining in the initial diabatic state π^* as a function of time. (d) Butatriene molecule with its electronic states represented by orbitals (pink and cyan) that differ between the two states, and the two vibrational modes Q_1 (red) and Q_2 (dark blue). (e) Molecular electronic and vibrational degrees of freedom can both be mapped onto a mixed qudit-boson (MQB) simulator consisting here of one trapped ion. Potentials and vibronic couplings are induced with lasers (yellow and green beams).

ity of the simulator by reproducing the real-time molecular dynamics of three different photoexcited molecules: the allene cation, the butatriene cation, and pyrazine. In each molecule, the vibronic couplings give rise to a conical intersection, which allows ultrafast (femtosecond) population transfer between electronic levels [17, 18]. Experimental measurements reliably reproduce the expected chemical dynamics, including signatures of the conical intersections. We further demonstrate the extensibility of our simulator to open-system dynamics by implementing simulations of pyrazine coupled to a thermal bath, showing the expected damping of coherent effects and thermalization. Our results further demonstrate the hardware efficiency of MQB simulation: with one trapped-ion qudit, two vibrational modes, and one laser pulse, it achieves a computation that would require 11 qubits and over 10^5 entangling gates in a standard digital quantum-computing framework.

I. MQB SIMULATION OF PHOTOCHEMISTRY

In an ion-trap MQB simulator, molecular vibrations and electronic states are encoded in motional and electronic degrees of freedom of the trapped ion (Fig. 1). We simulate non-adiabatic dynamics by reconstructing the time evolution of important molecular properties, rescaled to a timescale accessible by the simulator. Typ-

ically, temporal dynamics are rescaled from femtoseconds to milliseconds—i.e., by a factor of approximately 10^{11} —sufficient to enable direct probing with conventional electronics and laser systems.

Analog simulation of molecular dynamics comprises three stages (Fig. 2). First, the initial wavefunction is prepared by exciting the qudit and displacing the relevant motional modes. Second, using laser-ion interactions with frequencies and intensities chosen to reproduce the molecular Hamiltonian (calculated beforehand using electronic-structure theory), the simulator is evolved for some duration. Third, desired observables are measured. This process is repeated for varying evolution durations, allowing the reconstruction of observables as a function of time.

We simulate the photoinitiated non-adiabatic dynamics in three molecules: photoionized allene ($C_3H_4^+$), photoionized butatriene ($C_4H_4^+$), and photoexcited pyrazine ($C_4N_2H_4$). The three molecules exhibit a wide variety of photochemical dynamics due to differences in their potential energy surfaces (see Fig. 3a-c). Allene is photoionized to a degenerate pair of π states that are coupled via symmetry-breaking vibrational modes: bond alternation and torsion [33]. The potential is thus symmetric about the conical intersection along both vibrational modes. Butatriene is photoexcited to the cationic π^* state, coupled to the π state through central bond stretching and torsion [34, 35]. Both electronic

potential-energy surfaces are displaced along the bond-stretching mode, leading to a peaked conical intersection, i.e., one where the lower adiabatic surface decreases in energy in all directions away from the intersection. Finally, pyrazine is photoexcited to the bright (large transition dipole moment) $\pi\pi^*$ state and decays to the dark $n\pi^*$ state, and its dynamics is dominated by a ring breathing mode and an out-of-plane hydrogen wag [36, 37]. The resulting Hamiltonian leads to a sloped conical intersection, where the lower adiabatic potential energy increases in one direction (here, $-Q_1$).

Each of the three molecules is well described by a linear vibronic coupling (LVC) model, where the two electronic levels and the two vibrational modes are linearly coupled. Denoting the initially excited electronic state as $|1\rangle$ (and the other one as $|0\rangle$), the Hamiltonian is [24]

$$\hat{H}_{\text{mol}} = -\frac{1}{2}\Delta E \hat{\sigma}_z + \sum_j \omega_j \hat{a}_j^{\dagger} \hat{a}_j + \frac{\kappa}{\sqrt{2}} \hat{\sigma}_z (\hat{a}_1^{\dagger} + \hat{a}_1) + \frac{\lambda}{\sqrt{2}} \hat{\sigma}_x (\hat{a}_2^{\dagger} + \hat{a}_2), \quad (1)$$

where $\hat{\sigma}_{x,z}$ are the Pauli operators and \hat{a}_j is the annihilation operator of the jth mode with frequency ω_j . ΔE is the energy difference between the two electronic states, while κ and λ are the tuning and coupling parameters, respectively. The initial wavefunction is a coherent state displaced along the mode j=1 by $\hat{D}_1(\alpha)=\exp(\alpha\hat{a}_1^{\dagger}-\alpha^*\hat{a}_1)$. The parameter values for each molecule, taken from calculations in prior literature, are given in table 1.

II. TRAPPED-ION SIMULATION

Each term in $\hat{H}_{\rm mol}$ can be implemented on a trappedion MQB simulator consisting of a qubit (qudit with d=2) representing the electronic states of the molecule and two motional modes of the ion in its harmonic confining potential representing the vibrations (Fig. 1e) [24]. This nearly one-to-one correspondence between molecular and trapped-ion Hamiltonian terms underpins the straightforwardness of analog simulation. The vibronic couplings are implemented by tunable, laser-driven spin-dependent forces (SDF) in the Lamb-Dicke regime [24], which drive the motional modes depending on the state of the qubit. After interaction-frame transformations and rotating-wave approximations [24, 38], the Hamiltonian for interactions involving mode j is

$$\hat{H}_{j,\phi_s}^{\text{SDF}}(\delta,\Omega) = \frac{\Omega}{2} \left(\hat{\sigma}_x \cos \phi_s + \hat{\sigma}_y \sin \phi_s \right) \left(\hat{a}_j^{\dagger} + \hat{a}_j \right) + \delta \hat{a}_j^{\dagger} \hat{a}_j, \quad (2)$$

where $\hat{\sigma}_{x,y,z}$ are the Pauli matrices, ϕ_s is a phase that determines the qubit operator, which can be adjusted by the laser phase, Ω is the Rabi frequency that is tuneable with the laser intensity, and δ is the user-adjustable detuning of the laser frequency from the motional mode. We use the notation $\hat{H}_{j,x}^{\mathrm{SDF}}$ and $\hat{H}_{j,y}^{\mathrm{SDF}}$ for interactions where $\phi_s = 0$ and $\phi_s = \pi/2$, respectively. An SDF interaction in the $\hat{\sigma}_z$ basis can be obtained with a qubit basis rotation, $\hat{H}_{j,z}^{\mathrm{SDF}} = \hat{R}_x(\pi/2)\hat{H}_{j,y}^{\mathrm{SDF}}\hat{R}_x(-\pi/2)$, where $\hat{R}_x(\theta)$

	Allene [33]	Butatriene [35]	Pyrazine [37]
$ 1\rangle$	π_x	π^*	$\pi\pi^*$
$ 0\rangle$	π_y	π	$n\pi^*$
$F/10^{-11}$	1.08	1.10	1.33
$\omega_1/2\pi$ (THz)	22.5	62.9	17.9
$\omega_2/2\pi$ (THz)	57.3	22.0	28.5
$\Delta E/2\pi$ (THz)	0	131.5	199
$\kappa/2\pi$ (THz)	74.7	62.1	-30.7
$\lambda/2\pi$ (THz)	67.7	69.6	63.3
α	0	-0.140	0.210

Table 1. Parameters of the LVC Hamiltonian, through which the MQB simulator can be programmed to simulate different molecules. $|0\rangle$ and $|1\rangle$ are the two relevant electronic states of the molecule. ω_1 and ω_2 are the frequencies of the two vibrational modes. ΔE is the energy difference between the electronic states at the origin. κ and λ are the tuning and coupling parameters. α is the displacement of the initial wavepacket. Parameters programmed on the MQB simulator are obtained by scaling the molecular frequencies by F. All molecular parameters are obtained from electronic-structure calculations [33, 35, 37].

are laser-driven qubit rotations around the x-axis of the Bloch sphere. We can also implement a Hamiltonian that generates a qubit rotation around $\hat{\sigma}_z$ by sandwiching a laser-driven interaction $\hat{H}_y^{\rm Q}(\chi) = \chi \hat{\sigma}_y/2$ between two \hat{R}_x rotations, i.e., $\hat{H}_z^{\rm Q} = \hat{R}_x(\pi/2)\hat{H}_y^{\rm Q}\hat{R}_x(-\pi/2)$, where the Rabi frequency χ is tuneable with the laser's power.

 \hat{H}_{mol} is experimentally implemented by simultaneously driving two SDF interactions and a qubit rotation, resulting in the ion Hamiltonian $\hat{H}_{\mathrm{ion}} = F\hat{H}_{\mathrm{mol}}$, where a scaling factor F scales molecular frequencies (THz) and timescales (ps) to the trapped ion's frequencies (MHz) and timescales (ms). Overall,

$$\hat{H}_{\text{ion}} = \hat{H}_{1,z}^{\text{SDF}}(F\omega_1, \sqrt{2}F\kappa) + \hat{H}_{2,x}^{\text{SDF}}(F\omega_2, \sqrt{2}F\lambda) + \hat{H}_{z}^{\text{Q}}(-F\Delta E). \quad (3)$$

All parameters of eq. (3) (ω_1 , ω_2 , κ , λ , and ΔE) are programmable and can be set to simulate any LVC molecule. To obtain faster ion-trap dynamics, the scaling factor F is maximized within the constraint of the available laser power [24, 31].

A. Closed-System Experiment

We perform our experiment with a single $^{171}{\rm Yb}^+$ ion electromagnetically held in vacuum with a quadrupole ion trap [25, 26]. A combination of radiofrequency and static electric fields confine the ion and give rise to secular vibrational motions with frequencies $\{\omega_x,\omega_y,\omega_z\}=2\pi\times\{1.33,1.51,0.5\}$ MHz. These vibrations are harmonic to an excellent approximation. We encode the two molecular electronic states into the two magnetically insensitive hyperfine levels of the ion's $^2{\rm S}_{1/2}$ ground state, with labels $|0\rangle\equiv|F=0,m_F=0\rangle$ and $|1\rangle\equiv|F=1,m_F=0\rangle$. The molecular vibrational modes, B_1 and B_2 , are encoded in the vibrational modes along the radial x and y directions.

The experimental sequence for enabling the direct simulation of photochemical dynamics in the time domain

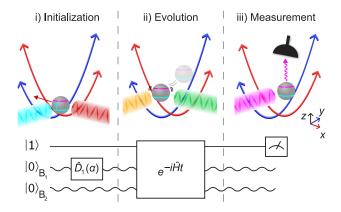


Figure 2. Experimental implementation of an MQB simulator for chemical dynamics. Three stages of MQB dynamics simulation: i) initialization, ii) evolution, and iii) measurement. Top row: trapped ion during each stage, with colors and symbols as in Fig. 1. Bottom row: circuit diagram of the pulse sequence acting on the qubit and two modes B_1 and B_2 (wavy lines). i) During initialization, the qubit is prepared in state $|1\rangle$ and the vibrational modes are prepared in their ground states. Then, B_1 is displaced by $\hat{D}_1(\alpha)$ using a state-dependent force enacted by laser beams (pink and red). ii) During evolution, the Hamiltonian is applied for duration t using state-dependent forces enacted by laser beams (yellow and green). iii) The electronic state is measured using a photon counter through state-dependent fluorescence (cyan).

is shown in Fig. 2 (methods are detailed in [25, 26]). We program the experiment with parameters of \hat{H}_{ion} chosen according to table 1 to implement each of the three target molecules. In each case, the scaling factor F is chosen to be $F = \Omega_1/\sqrt{2}\kappa$, where Ω_1 is the Rabi frequency of the SDF interaction of eq. (2) with the mode B_1 by setting j = 1. The specific implementation of each simulation follows the three-stage process introduced above. First, initialization prepares the qubit in $|1\rangle$ and the modes in their ground states, followed by displacing B_1 by $D_1(\alpha)$. Second, the system is evolved by applying \hat{H}_{ion} for an experimentally variable duration $t_{\rm ion}$, which is related to the molecular timescale by $t_{\text{mol}} = Ft_{\text{ion}}$. Third, measurement of the electronic populations is achieved by making a $\hat{\sigma}_z$ measurement on the qubit through state-dependent fluorescence [39].

Experimental measurements of the probability of finding the wavepacket in the initial diabatic state as a function of evolution time are shown in Fig 3d-f. For all three molecules, experiments match the predicted dynamics arising from conventional calculations. Small discrepancies are most likely due to miscalibrated experimental parameters and hardware noise. In all cases, we observe a population decay due to the rapid non-radiative decay of the wavepacket through the conical intersection. We further observe distinct behaviors for each molecule, caused by the different potential-energy surfaces. The symmetric conical intersection of the allene cation causes a rapid decay of the population because the initial state is at the point of strongest coupling between electronic states; however, only a small amount of population is transferred due to the degeneracy of the π states. The butatriene cation likewise has an initial state with strong coupling, but the peaked conical intersection leads to

roughly half the population remaining on the π^* state. Pyrazine has a sloped conical intersection that leads to a near-complete occupation of the lower $n\pi^*$ state, but the initial dynamics are much slower than the other two molecules.

B. Open-System Experiment

To further demonstrate the versatility of the MQB approach, we perform a quantum simulation of open-system vibronic dynamics, which enables the simulation of environmental molecular conditions. We build on the closed-system simulation above, and externally inject noise with the same single trapped ion. Doing so shows that what is classically a harder computational problem can be solved using the same quantum resources on an MQB simulator [31].

In particular, we simulate the non-equilibrium ultrafast dynamics of a pyrazine molecule interacting with a thermal bath. The dissipation is simulated by simultaneously injecting heating and cooling of the vibrational modes, described by the Lindblad master equation [40, 41],

$$\frac{d\hat{\rho}}{dt} = -i[\hat{H}_{\text{ion}}, \hat{\rho}] + \sum_{j} \left(\gamma_{+,j} \mathcal{D}[\hat{a}_{j}^{\dagger}] \hat{\rho} + \gamma_{-,j} \mathcal{D}[\hat{a}_{j}] \hat{\rho} \right), (4)$$

where $\mathcal{D}[\hat{a}_{j}^{\dagger}]$ is the heating dissipator for mode j,

$$\mathcal{D}[\hat{a}_j^{\dagger}]\hat{\rho} = \hat{a}_j^{\dagger}\hat{\rho}\hat{a} - \frac{1}{2}\{\hat{a}_j^{\dagger}\hat{a}, \hat{\rho}\},\tag{5}$$

and similarly for the cooling dissipator $\mathcal{D}[\hat{a}]\hat{\rho}$. Heating and cooling have rates $\gamma_{+,j}$ and $\gamma_{-,j}$, respectively, and describe energy transfer from the environment to the vibrations (heating) or vice versa (cooling).

The dissipators are engineered in the simulator by injecting a noisy electric field at the ion's location [42]. The electric field noise results in cooling and heating dissipators with approximately equal rates, $\gamma_j = \gamma_{+,j} = \gamma_{-,j}$, corresponding to a thermal bath with infinite temperature [40, 41]. In the molecule, this corresponds to a scaled dissipation rate $\Gamma_i = F^{-1}\gamma_i$. A noisy voltage signal, generated using an arbitrary waveform generator, is capacitively coupled onto a radial compensation electrode located 4.8 mm from the ion's position. The signal contains two frequency components oscillating near the vibrational mode frequencies, ω_x and ω_y . The rates γ_1 and γ_2 are programmed by varying the corresponding signal amplitudes. The decay rates are calibrated using the trapped ion through a standard sideband thermometry experiment [43, 44].

The experiment correctly simulates the open-system dynamics at varying dissipation rates (Fig. 3f), showing the suitability of MQB simulation for the particularly challenging task of modelling ultrafast dynamics in open molecular systems. As expected, stronger dissipation leads to a faster loss of coherent features. With dissipation, the electronic population decays at long times to 0.5, consistent with a Boltzmann distribution at high temperature, where the populations of the two electronic states obey $n_1/n_0 = \exp(-\Delta E/k_{\rm B}T) = 1$. This equilibration occurs faster with stronger dissipation.

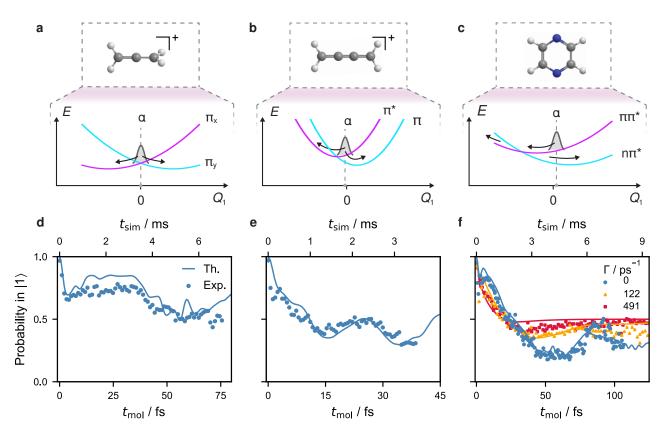


Figure 3. Quantum simulation of photochemical dynamics in (a,d) photoionized allene, (b,e) photoionized butatriene, and (c,f) photoexcited pyrazine. (a–c) Adiabatic potential energy surfaces of each molecule (one-dimensional slice at $Q_2 = 0$), with the wavepacket (gaussian width not to scale) initialized by displacement to $Q_1 = \alpha$. Each molecule results in a distinct energy landscape, where the conical intersection is (a) symmetric, (b) peaked, and (c) sloped. Black arrows indicate the possible dynamics pathways for wavepackets to evolve on the PESs. All molecules were simulated on an ion-trap MQB simulator using the LVC Hamiltonian of eq. (3) with parameters from table 1. (d–f) The probability of finding the wavepacket in the initial diabatic state as a function of the molecular evolution time (bottom axes). Additionally, in (f), open-system dynamics of pyrazine under heating and cooling dissipation with varying rates Γ . Solid lines are the solutions of the Schrödinger equation for closed-system dynamics and of the master equation (eq. (4)) for open-system dynamics. The simulator times shown in the top axes correspond to the closed-system dynamics. The scaling factors for the open-system dynamics with $\Gamma = \{122, 491\}$ ps⁻¹ are $F = \{1.69, 1.24\} \times 10^{-11}$. Experimental data points are obtained by averaging over 500 measurement results. Error bars corresponding to quantum projection noise are smaller than the plot markers.

III. DISCUSSION

In summary, we performed quantum simulations of non-adiabatic chemical dynamics of three molecules. We simulated the evolution of their vibrational wavepackets through conical intersections, obtaining distinct dynamics linked to the characteristics of their potential energy surfaces. We also performed open-system dynamics simulations of pyrazine coupled to a thermal bath using the same trapped-ion system.

Our experiment demonstrates three advantages of analog MQB simulations: programmability, resource efficiency, and the ability to simulate open systems.

MQB simulators are programmable, allowing us to simulate the dynamics of very different molecules simply by adjusting parameters in the same experiment. This programming is carried out through software-updated experimental parameters, without the need for hardware modification. Different features of photochemical processes were captured by the same MQB simulator: dynamics between different types of electronic states (ground or excited, singlet or doublet, degenerate or non-degenerate), vibrational modes and vibronic couplings

with different symmetries, conical intersections with different topography (symmetric, peaked, and sloped), and a wide range of vibrational frequencies, vibronic couplings and wave-packet displacements.

MQB simulators require orders of magnitude less quantum-hardware resources compared to digital quantum computing approaches for chemical dynamics simulations [12, 24]. We used a single trapped ion and a single laser pulse to simulate the ultrafast molecular dynamics of LVC models with two electronic states and two vibrational modes. The comparable digital quantum memory requirements can be estimated from the dimension of the necessary Hilbert space. Our simulations can be reproduced using 32 Fock states per mode, meaning that a comparable digital simulation would need 11 qubits: $\lceil \log_2 32 \rceil = 5$ qubits for each of the two modes and an additional qubit for the electronic states. The digital gate requirements can be estimated using the resource-efficient Gray-code quditto-qubit encoding [45] with accuracy chosen to reproduce the mean-squared error of our MQB simulation (which is 0.0034, averaged over the duration of our simulation). Doing so with a first-order Suzuki-Trotter decomposition

needs 1000 errorless CNOT gates for each of 300 Trotter steps. Therefore, a comparable digital simulation could be achieved with 3×10^5 CNOT gates. A realistic digital quantum computer with noisy gates or quantum error correction would require even greater digital resources.

Finally, we confirmed the ability of MQB simulators to simulate open-system dynamics by injecting controllable noise without using additional quantum resources. Comparable simulations on digital quantum computers would have required additional ancilla resources to simulate the environment. On the MQB simulator, by contrast, the classically more difficult open-system problem is an easier task because some of the native noise can be used in the simulation, allowing for longer simulations with greater accuracy [31].

Extending MQB simulation beyond proof-of-principle experiments will enable increasingly challenging dynamics simulations. As proposed earlier, MQB simulators could be extended to implement higher-order vibronic-coupling models [24] or anharmonic potential energy surfaces. Moreover, with the same trapped-ion system, one could include other types of dissipation for more comprehensive and fully programmable open-system quantum dynamics [31]. Combining these improvements would allow addressing one of the great challenges of computational chemistry: simulation of the ultrafast dynamics of large, complicated molecules in the condensed

phase. In doing so, MQB simulators could outperform other quantum simulation approaches and offer a pathway to achieving quantum advantage in the near term. We estimate that an MQB simulator with 20–30 trapped ions could perform full quantum chemical-dynamics simulations that are impossible classically.

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DATA AVAILABILITY

Experimental data are available upon reasonable request.

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- [1] R. P. Feynman, Int. J. Theor. Phys. 21, 467–488 (1982).
- [2] S. Lloyd, Science **273**, 1073–1078 (1996).
- [3] B. P. Lanyon, J. D. Whitfield, G. G. Gillett, M. E. Goggin, M. P. Almeida, I. Kassal, J. D. Biamonte, M. Mohseni, B. J. Powell, M. Barbieri, A. Aspuru-Guzik, and A. G. White, Nat. Chem. 2, 106–111 (2010).
- [4] Y. Cao, J. Romero, J. P. Olson, M. Degroote, P. D. Johnson, M. Kieferová, I. D. Kivlichan, T. Menke, B. Peropadre, N. P. D. Sawaya, S. Sim, L. Veis, and A. Aspuru-Guzik, Chem. Rev. 119, 10856–10915 (2019).
- [5] S. McArdle, S. Endo, A. Aspuru-Guzik, S. C. Benjamin, and X. Yuan, Rev. Mod. Phys. 92, 015003 (2020).
- [6] A. Aspuru-Guzik, A. D. Dutoi, P. J. Love, and M. Head-Gordon, Science 309, 1704–1707 (2005).
- [7] A. Peruzzo, J. McClean, P. Shadbolt, M.-H. Yung, X.-Q. Zhou, P. J. Love, A. Aspuru-Guzik, and J. L. O'Brien, Nat. Comm. 5, 1 (2014).
- [8] A. Kandala, A. Mezzacapo, K. Temme, M. Takita, M. Brink, J. M. Chow, and J. M. Gambetta, Nature 549, 242–246 (2017).
- [9] C. Hempel, C. Maier, J. Romero, J. McClean, T. Monz, H. Shen, P. Jurcevic, B. P. Lanyon, P. Love, R. Babbush, A. Aspuru-Guzik, R. Blatt, and C. F. Roos, Phys. Rev. X 8, 3 (2018).
- [10] B. Bauer, S. Bravyi, M. Motta, and G. K.-L. Chan, Chem. Rev. 120, 12685–12717 (2020).
- [11] Y. Nam, J.-S. Chen, N. C. Pisenti, K. Wright, C. Delaney, D. Maslov, K. R. Brown, S. Allen, J. M. Amini, J. Apisdorf, K. M. Beck, A. Blinov, V. Chaplin, M. Chmielewski, C. Collins, S. Debnath, K. M. Hudek, A. M. Ducore, M. Keesan, S. M. Kreikemeier, J. Mizrahi, P. Solomon, M. Williams, J. D. Wong-

- Campos, D. Moehring, C. Monroe, and J. Kim, npj Quantum Inf. 6, 1 (2020).
- [12] I. Kassal, S. P. Jordan, P. J. Love, M. Mohseni, and A. Aspuru-Guzik, Proc. Natl. Acad. Sci. 105, 18681–18686 (2008).
- [13] N. P. D. Sawaya and J. Huh, J. Phys. Chem. Lett. 10, 3586–3591 (2019).
- [14] P. J. Ollitrault, G. Mazzola, and I. Tavernelli, Phys. Rev. Lett. 125, 26 (2020).
- [15] A. Miessen, P. J. Ollitrault, F. Tacchino, and I. Tavernelli, Nat. Comput. Sci. 3, 25 (2023).
- [16] M. D. E. Forbes, ACS Cent. Sci. 1, 354–363 (2015).
- [17] W. Domcke, D. R. Yarkony, and H. Köppel, Conical Intersections: Electronic Structure, Dynamics and Spectroscopy (World Scientific, 2004).
- [18] L. S. Cederbaum, H. Köppel, and W. Domcke, Int. J. Quantum Chem. 20, 251 (1981).
- [19] G. Worth, H. Meyer, H. Koeppel, L. Cederbaum, and I. Burghardt, Int. Rev. Phys. Chem. 27, 569 (2008).
- [20] A. Nitzan, Chemical Dynamics in Condensed Phases (Oxford University Press, 2006).
- [21] G. A. Worth, H.-D. Meyer, and L. S. Cederbaum, J. Chem. Phys. 109, 3518 (1998).
- [22] M. Bonfanti, G. F. Tantardini, K. H. Hughes, R. Martinazzo, and I. Burghardt, J. Phys. Chem. A 116, 11406 (2012).
- [23] A. V. Haeften, C. Ash, and G. Worth, J. Chem. Phys. 159, 194114 (2023).
- [24] R. J. MacDonell, C. E. Dickerson, C. J. T. Birch, A. Kumar, C. L. Edmunds, M. J. Biercuk, C. Hempel, and I. Kassal, Chem. Sci. 12, 9794–9805 (2021).
- [25] R. J. MacDonell, T. Navickas, T. F. Wohlers-Reichel, C. H. Valahu, A. D. Rao, M. J. Millican, M. A. Currington, M. J. Biercuk, T. R. Tan, C. Hempel, and

- I. Kassal, Chem. Sci. 14, 9439–9451 (2023).
- [26] C. H. Valahu, V. C. Olaya-Agudelo, R. J. MacDonell, T. Navickas, A. D. Rao, M. J. Millican, J. B. Pérez-Sánchez, J. Yuen-Zhou, M. J. Biercuk, C. Hempel, T. R. Tan, and I. Kassal, Nat. Chem. 15, 1503-1508 (2023).
- [27] J. Whitlow, Z. Jia, Y. Wang, C. Fang, J. Kim, and K. R. Brown, Nat. Chem. 15, 1509–1514 (2023).
- [28] V. So, M. D. Suganthi, A. Menon, M. Zhu, R. Zhuravel, H. Pu, P. G. Wolynes, J. N. Onuchic, and G. Pagano, "Trapped-ion quantum simulation of electron transfer models with tunable dissipation," (2024), arXiv:2405.10368.
- [29] M. Kang, H. Nuomin, S. N. Chowdhury, J. L. Yuly, K. Sun, J. Whitlow, J. Valdiviezo, Z. Zhang, P. Zhang, D. N. Beratan, and K. R. Brown, Nat. Rev. Chem. 8, 340–358 (2024).
- [30] K. Sun, M. Kang, H. Nuomin, G. Schwartz, D. N. Beratan, K. R. Brown, and J. Kim, "Quantum simulation of spin-boson models with structured bath," (2024), arXiv:2405.14624.
- [31] V. C. Olaya-Agudelo, B. Stewart, C. H. Valahu, R. J. MacDonell, M. J. Millican, V. G. Matsos, F. Scuccimarra, T. R. Tan, and I. Kassal, "Simulating open-system molecular dynamics on analog quantum computers," (2024), arXiv:2407.17819.
- [32] Y. Shen, Y. Lu, K. Zhang, J. Zhang, S. Zhang, J. Huh, and K. Kim, Chemical Science 9, 836–840 (2018).
- [33] C. Woywod and W. Domcke, Chem. Phys. 162, 349 (1992).
- [34] C. Cattarius, G. A. Worth, H.-D. Meyer, and L. S. Cederbaum, J. Chem. Phys. 115, 2088 (2001).

- [35] I. G. Ryabinkin, L. Joubert-Doriol, and A. F. Izmaylov, J. Chem. Phys. 140, 214116 (2014).
- [36] C. Woywod, W. Domcke, A. L. Sobolewski, and H.-J. Werner, J. Chem. Phys. 100, 1400 (1994).
- [37] A. Kühl and W. Domcke, J. Chem. Phys. 116, 263 (2002).
- [38] D. J. Wineland, C. Monroe, W. M. Itano, D. Leibfried, B. E. King, and D. M. Meekhof, J. Res. Natl. Inst. Stand. Technol. 103, 259 (1998).
- [39] S. Olmschenk, K. C. Younge, D. L. Moehring, D. N. Matsukevich, P. Maunz, and C. Monroe, Phys. Rev. A 76, 5 (2007).
- [40] C. Henkel, S. Pötting, and M. Wilkens, App. Phys. B 69, 379–387 (1999).
- [41] M. Brownnutt, M. Kumph, P. Rabl, and R. Blatt, Rev. Mod. Phys. 87, 1419–1482 (2015).
- [42] C. H. Valahu, T. Navickas, M. J. Biercuk, and T. R. Tan, "Benchmarking bosonic modes for quantum information with randomized displacements," (2024), arXiv:2405.15237.
- [43] C. Monroe, D. M. Meekhof, B. E. King, S. R. Jefferts, W. M. Itano, D. J. Wineland, and P. Gould, Phys. Rev. Lett. 75, 4011 (1995).
- [44] Q. A. Turchette, Kielpinski, B. E. King, D. Leibfried, D. M. Meekhof, C. J. Myatt, M. A. Rowe, C. A. Sackett, C. S. Wood, W. M. Itano, C. Monroe, and D. J. Wineland, Phys. Rev. A 61, 6 (2000).
- [45] N. P. D. Sawaya, T. Menke, T. H. Kyaw, S. Johri, A. Aspuru-Guzik, and G. G. Guerreschi, npj Quantum Inf. 6, 49 (2020).